

$v_f = v_0 + at$ $x = (1/2)(v_0 + v_f)t$ $x = v_0t + (1/2)at^2$ $v_f^2 = v_0^2 + 2ax$	$\sigma = s/r \quad \omega = v/r \quad \alpha = a_t/r$ $\varpi = \varpi_0 + \alpha t$ $\sigma = (1/2)(\varpi_0 + \varpi)t$ $\sigma = \varpi_0t + (1/2)\alpha t^2$ $\varpi^2 = \varpi_0^2 + 2\alpha\sigma$
$F=ma$	$\tau = I \alpha$
$p=mv$	$L = I\omega$
$W=F\cos\theta\Delta s$	$W=\tau\sigma$

the force of gravity $F=Gm_1m_2/r^2$ where $G=6.67 \times 10^{-11} \text{ N m}^2/\text{kg}^2$

rotational speed = $2\pi r/T$

centripetal acceleration = $v^2/r = \omega^2 r$

circumference = $2\pi r$

$F\Delta t = mv_f - mv_i$ $(1/2)mv_i^2 + mgh_i =$ $(1/2)mv_f^2 + mgh_f$ K.E. = $0.5 mv^2$ (no rotation) K.E. = $0.5 mv^2 + 0.5 I \varpi^2$ (with rotation) $\tau = r F$ $I = mr^2$ angular momentum = $L = I\omega$	PE(spring) = $0.5kx^2$ F(spring) = $-kx$ $\rho = \text{mass}/V \quad V = Ax \text{ (or Ah)}$ P = F/A $A_1v_2 = A_2v_1$ $P_2 = P_1 + \rho g\Delta h$ $F_B = \text{weight of displaced fluid}$ Bernoulli: $P_1 + 0.5\rho v_1^2 + \rho gh_1 = P_2 + 0.5\rho v_2^2 + \rho gh_2$
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273K = 0°C

atmospheric pressure = 10^5 N/m^2

density lead $11,300 \text{ kg/m}^3$

density water $1,000 \text{ kg/m}^3$; ice 917 kg/m^3

density mercury $13,600 \text{ kg/m}^3$

density pine 550 kg/m^3

density air 1.29 kg/m^3

latent heat of fusion water $3.35 \times 10^5 \text{ J/kg}$

latent heat of vaporization water $22.6 \times 10^5 \text{ J/kg}$

heat capacity water = 4186 J/Kg/°

ice = 2000 J/Kg/°

water vapor = 2020

lead = 128 J/kg/°

$Q=mc T$

$Q=mL_f$ or mL_v

1 mole of ideal gas at
0°C and 1atm pressure
vol= $22.4 \times 10^{-3} \text{ m}^3$

constant pressure work = $P V$

$U = Q_{in} - W_{out}$

$Q_{Hot} = W + Q_{cold} \quad Q_{cold}/Q_{hot} = T_{cold}/T_{hot}$

eff= $W/Q_{hot} = 1 - Q_{cold}/Q_{hot} = 1 - T_{cold}/T_{hot}$

ideal gas

$PV = nRT = NkT$

internal energy = $U = 3/2 NkT = 3/2 nRT$

Av KE = $1/2 mv^2 = 3/2 kT$ (av molec velocity)

$k = 1.38 \times 10^{-23} \text{ J/K}$

$R = 8.31 \text{ J/(mol K)}$

$N = 6.02 \times 10^{23} / \text{mol}$

Molecular weight

$O_2 = 32 \text{ g/mole}$

$N_2 = 28$

Argon = 40